

Appl. No 10/743, 641
Amtd. Dated December 27, 2007
Reply to Office action of June 27, 2007

ARGUMENTS/REMARKS

This communication is responsive to the Office Action dated June 27, 2007.

Claims 1-55 remain pending.

The Office Action

In the Office Action of June 27, 2007, the Examiner made the following rejections:

Claims 2-9, 11-18, 20, 21, 31, 34-36, 38, 42-44 and 50 were rejected under 35 U.S.C. §112, second paragraph, as being indefinite.

Claims 1, 4, 5, 11-13, 16 and 34 were rejected under 35 U.S.C. §102 (b), as being anticipated by, or under 35 USC § 103(a) as being obvious over, Hirano et al (U.S. Patent No. 5,028,495).

Claims 1, 4, 5, 11-13, 16 and 34 were rejected under 35 U.S.C. §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Jha et al (U.S. Patent No. 5,553,770).

Claims 1-3, 7, 8, 10-13, 16, 19, 20, 22, 25, 26, 29 and 34 were rejected under 35 U.S.C. §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Mennucci (U.S. Patent No. 5,761,799).

Claims 1, 4, 5, 10, 11-16 and 34 were rejected under 35 U.S.C. §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Galasso et al (U.S. Patent No. 4,034,454).

Claims 1-30, 34-46, 49, 50 and 53-55 were rejected under 35 U.S.C. §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Hirano et al (JP 4-006173).

Claims 32, 33, 47, 48, 51 and 52 were objected to as being dependent upon a rejected base claim but otherwise identified as being allowable.

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Claims 31 was identified as being allowable if amended to overcome the rejection under 35 U.S.C. §112, second paragraph.

The Rejections Under 35 U.S.C §112

Claims 2-9, 11-18, 20, 21, 31, 34-46, 38, 42-44 and 50 were rejected under 35 U.S.C. §112, second paragraph, as the examiner maintains that the term "commercially pure" is unclear since it can change over time.

In the prior response, Applicants submitted that the materials used in the invention are to be as pure as whatever is commercially available. Should higher purity materials become available than what was known at the time of the invention, then those materials would be preferred over what was commercially available at the present time and would work just as well as the presently available commercial materials.

In addition to the above, and as pointed out in the previously filed response, Applicants would like to note that a quick search on the USPTO's online database revealed at least 143 issued U.S. patents with the term "commercially pure" in the claims. This includes U.S. Patent No. 6,783,870, assigned to the applicant of the present application (Engineered Material Solutions). See attached listing of issued U.S. patents which contain the term "commercially pure" in the claims (Exhibit "A").

Further, an internet search using "Google" revealed a number of technical reports and technical dictionary definitions of the term "commercially pure" in relation to metals such as copper and nickel.

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Again, the intent is to use material which has a purity level that is as high as commercially available. Applicants believe that this is clear and definite within the meaning of 35 U.S.C. §112, second paragraph and supported by the *issuance of a number of U.S. patents* with the exact same language, including one to the present applicant.

In view of the above, withdrawal of the rejection under 35 U.S.C. §112, second paragraph, is respectfully requested.

Claim 42 was rejected under 35 U.S.C §112, second paragraph, as the examiner indicated that it was unclear what was meant by "is made from one of a group consisting of copper and nickel".

Applicants submit that the terminology is clear. When, for example, the first metallic layer is "copper", then the third metallic layer is "nickel" (which is the "other" of the two not used for the first layer). Likewise with the material for the forth and sixth layers. Applicants believe this is clear and definite. However, in an attempt to address the Examiner's concern, Applicants have now amended claim 42 in a manner similar to that suggested by the Examiner. The amended claim now requires that when the first layer is made from a metal selected from the group consisting of copper or nickel then the third layer is made from the other metal selected from the group consisting of copper or nickel. The same amendment was made for layers four and six in claim 42.

Applicants believe that the above comments and amendments fully address the rejections under 35 U.S.C §112, second paragraph. Reconsideration and withdrawal of the rejections are therefore respectfully requested.

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The Rejections Under 35 U.S.C §102(b)

Claims 1, 4, 5, 11-13, 16 and 34 were rejected under 35 U.S.C. §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Hirano et al (U.S. Patent No. 5,028,495). Further, claims 1, 4, 5, 11-13, 16 and 34 were rejected under 35 U.S.C. §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Jha et al (U.S. Patent No. 5,553,770). Also, claims 1, 4, 5, 10, 11-16 and 34 were rejected under 35 U.S.C. §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Galasso et al (U.S. Patent No. 4,034,454).

Applicants would like to address the above three references of Hirano et al, Jha et al and Galasso et al as all three references require a heat treatment step of the material. The products of the invention which result from a roll bonding process without any intermediate heat treatment step are not the same as those produced from processes which utilize a heating step.

Applicants would like to emphasize that the materials used in the present invention, in particular Ti and/or Zr, possess unique characteristics and qualities which limit the way in which they can be processed into useful materials. The prior art cited by the Examiner clearly fails to recognize these unique characteristics and the drawbacks which are realized by processing these materials in a manner requiring intermediate heat treatment steps, such as annealing.

Specifically, roll bonding of metal layers to Ti (and/or Zr) requires cold working and thus hardens the strip. As disclosed in the specification, roll bonding is typically carried out in a bonding mill that provides sufficient pressure to form a metallurgical bond without heat treating (such as annealing, for example) between dissimilar metals. It is therefore common metal working practice to put such cold worked strips through an annealing process to soften it up for any further cold rolling. This is what the prior art cited by the Examiner does. The annealing requires time at elevated

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temperatures and usually requires a protective atmosphere to avoid forming surface scales. The annealing process conditions that are likely to apply to the strip have two problems that will cause strip to become brittle. The resulting strip is then not able to be further cold rolled to the foil gauge for brazing filler application due to its brittleness. Further, the time and temperature will cause interdiffusion between Ti/Zr and Cu/Ni layers. The resultant intermetallic compounds between Ti/Zr and Cu/Ni are extremely brittle and will crack between layers under the cold rolling condition. A second issue is the gas used as the protective atmosphere. The commonly used gas is either nitrogen or hydrogen or a mixture of nitrogen and hydrogen, which will be absorbed into the Ti/Zr and cause further embrittlement. This will also render the strip even more unworkable. In summary, there are many metallurgical reasons that clad strips of the invention not be processed with conventional rolling and annealing practices as taught in the cited art. On the other hand, by cladding layers to the surfaces of Ti or Zr without an intermediate heating step, the ability to roll the clad strip to far more thickness reduction than what is possible on the Ti or Zr by itself or when clad with materials using an intermediate heating step is achieved. This is **completely unexpected and not taught or suggested by the cited art.**

Moreover, the Examiner is ignoring the limitation that the final "brazing strip or foil" is complete upon roll bonding without need for further treatment as instructed by the cited art. As such, it is our position that the prior art actually teaches away from the invention by including additional heat treatment steps and not recognizing that such steps are not required or encompassed by the current claims to produce the claimed brazing strip/foil.

In view of the above, the references of Hirano et al, Jha et al and Galasso et al do not teach the material of the invention as the heating steps used to prepare these products result in a clad

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material which is not the same as the claimed material which is made by a process which does not include any intermediate heating. As such, Hirano et al, Jha et al and Galasso et al neither anticipate nor obviate the designated claims under 35 USC § 102(b) or 103(a) and the rejections thereover should be withdrawn.

Claims 1-3, 7, 8, 10, 11-13, 16, 19, 20, 22, 25, 26, 29 and 34 were rejected under 35 U.S.C §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Mennucci (U.S. Patent No. 5,761,799).

The process of Mennucci et al required inclusion of platinum stripes (item no 15) to the surface of a copper layer which is to be bonded to a titanium base material. No such platinum stripes are present in the material of the invention. As such, the material of Mennucci et al is not the same as the material of the invention. Moreover, the product of Mennucci is not a brazing strip or foil but rather a laminate which is essentially a platinum coated titanium strip useful as anodes in chemical processing or in electrical conductor applications. No mention is made of using the platinum coated materials as brazing strips or foils. In fact, alloying of platinum (Pt) is needed in order to bring the Pt to a melting point necessary for brazing operations. Without such alloying of the Pt, it will not function as a brazing material and will remain as a discreet layer. Mennucci does not alloy the Pt as the Pt materials used therein are not intended for use as a brazing material and does not need to possess the melting properties required by brazing materials. As such, Mennucci fails to anticipate the claims under 35 U.S.C. §102 (b) or render the claims obvious under 35 USC § 103(a). Withdrawal of the rejection is respectfully requested.

Claims 1-30, 34-45, 48, 49 and 52-54 were rejected under 35 U.S.C §102 (b) as being anticipated by, or under 35 USC § 103(a) as being obvious over, Hirano et al (JP 4-006173).

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With respect to Hirano et al (JP), no mention is made of any metallic bond being present between layers, muchless that the bond was formed solely by roll bonding the materials together without any intermediate heating step. As discussed above, materials of the invention which contain a Ti or Zr core material which are produced without the intermediate heating are materially different from those produced in a process which utilizes some heating. While Hirano teaches "coating" a Zr base with a special Cu/Ti/Ni-Cu composite, rolling and annealing are required as seen in the translation at pages 6 and 7. Again, the claims of the present invention specifically require the absence of any intermediate heat treating (i.e. annealing is a heat treatment) prior to forming the final brazing strips or foils. Thus a reference which teaches some heating step prior to formation of the finished product does exactly what the claims exclude.

Moreover, the claims of the present invention require only one core, either Zr, Ti or a blend of Zr and Ti. The material of Hirano et al has a Zr core and then a material having a second Ti core (the Cu/Ti/Ni-Cu) material. This is not the same as the invention. As such, Claims 1-30, 34-45, 48, 49 and 52-54 are not anticipated by or obvious over the Hirano et al (JP) reference. Withdrawal of the rejection is therefore respectfully requested.

Conclusion

In consideration of the foregoing analysis, it is respectfully submitted that the present application is in a condition for allowance and notice to that effect is hereby requested. If it is determined that the application is not in a condition for allowance, the examiner is invited to initiate a telephone interview with the undersigned attorney to expedite prosecution of the present application.

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Respectfully submitted,

PEARNE & GORDON LLP

By:


Brian G. Bembenick, Reg. No. 41463

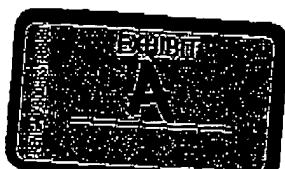
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PAT. NO.	Title
1 7,311,873	Process of direct powder rolling of blended titanium alloys, titanium matrix composites, and titanium aluminides
2 7,261,914	Method and apparatus for forming a nitride layer on a biomedical device
3 7,241,328	Method for preparing ultra-fine, submicron grain titanium and titanium-alloy articles and articles prepared thereby
4 7,205,179	Hydrogen diffusion hybrid port and method of making
5 7,201,210	Casting of aluminum based wrought alloys and aluminum based casting alloys
6 7,174,212	Implantable medical device having a casing providing high-speed telemetry
7 7,137,436	Method and apparatus for melting titanium using a combination of plasma torches and direct arc electrodes
8 7,048,590	High performance, high capacitance gain, jack connector for data transmission or the like
9 7,022,911	Apparatus and method for providing electrostatic discharge protection
10 7,009,176	Titanium ion transfer components for use in mass spectrometry
11 6,964,587	High performance, high capacitance gain, jack connector for data transmission or the like
12 6,919,623	Hydrogen diffusion hybrid port and method of forming
13 6,814,820	Heat treatment of titanium-alloy article having martensitic structure
14 6,797,662	Electrically conductive ceramics
15 6,797,006	Porous unicondylar knee
16 6,783,870	Self-brazing materials for elevated temperature applications
17 6,763,524	Titanium wire face guard



18 6,749,610 **Electro-surgical forceps having fully plated tines and process for manufacturing same**
19 6,740,186 **Method of making an orthopedic implant having a porous metal surface**
20 6,731,849 **Coating for optical fibers**
21 6,638,381 **Method for preparing ultra-fine grain titanium and titanium-alloy articles and articles prepared thereby**
22 6,637,091 **Method for making titanium wire face guard**
23 6,610,962 **Method for producing a crack resistant weld**
24 6,591,636 **Material and method for coating glass forming equipment**
25 6,553,097 **X-ray tube anode assembly and x-ray systems incorporating same**
26 6,543,273 **Efficient use of metallic materials for dynamic tear testing**
27 6,542,566 **Cladding for use in nuclear reactors having improved resistance to stress corrosion cracking and corrosion**
28 6,541,735 **Bearing shaft assembly having a crack resistant weld**
29 6,517,801 **Treatment of gas streams containing hydrogen sulphide**
30 6,419,491 **Dental implant system with repeating microgeometric surface patterns**
31 6,410,165 **Crack resistant weld**
32 6,335,512 **X-ray device comprising a crack resistant weld**
33 6,325,805 **Shape memory alloy staple**
34 6,322,364 **Superplastically-formed prosthetic components, and equipment for same**
35 6,312,473 **Orthopedic implant system**
36 6,273,990 **Method and apparatus for removing a protective coating from an optical fiber and inhibiting damage to same**
37 6,248,105 **Device for connecting a longitudinal support with a pedicle screw**
38 6,243,433 **Cladding for use in nuclear reactors having improved resistance to stress corrosion cracking and corrosion**
39 6,210,807 **Surface oxidation of a titanium or titanium alloy article**
40 6,199,259 **Autoclave bonding of sputtering target assembly**
41 6,123,899 **Master alloy hardeners**
42 6,116,070 **Superplastically-formed prosthetic components, and equipment for same**
43 6,102,954 **Joint prosthesis and apparatus for preparing the bone prior to fitting of the prosthesis**
44 6,048,204 **Self tapping screw type dental implant**
45 6,021,155 **Heat treating furnace having improved hot zone**
46 5,961,329 **Combination distraction dental implant and method of use**
47 5,868,875 **Non-ridging ferritic chromium alloyed steel and method of making**
48 5,843,289 **Surface modification of medical implants**
49 5,814,048 **Cranioplasty plates and method of installation**
50 5,812,925 **Low temperature bonding of materials**

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PAT. NO. Title

51 [5,800,895](#) **Beryllium memory disk substrate for computer hard disk drive and process for making**

52 [5,769,637](#) **Dental implant and alveolar process augmentation structures and method of installation**

53 [5,744,254](#) **Composite materials including metallic matrix composite reinforcements**

54 [5,732,767](#) **Corrosion resistant heat exchanger and method of making the same**

55 [5,527,995](#) **Cable for conducting energy**

56 [5,486,340](#) **Exhaust gas conditioning**

57 [5,443,510](#) **Porous coated implant and method of making same**

58 [5,442,929](#) **Cryogenically-treated electrical contacts**

59 [5,435,375](#) **Titanium composite casting nozzle**

60 [5,429,780](#) **Manufacture of silicon carbide foam from a polyurethane foam impregnated with resin containing silicon**

61 [5,423,820](#) **Surgical cable and crimp**

62 [5,418,072](#) **Totally consumable brazing encapsulate for use in joining aluminum surfaces**

63 [5,405,578](#) **Method for preparing master alloy hardeners for use in preparing an aluminum alloy**

64 [5,399,376](#) **Meniscus coating steel strip**

65 [5,371,178](#) **Rapidly curing adhesive and method**

66 [5,332,454](#) **Titanium or titanium based alloy corrosion resistant tubing from welded stock**

67 5,330,587 **Shaft of laser nitride-hardened surface on titanium**
68 5,326,851 **Low-viscosity polyarylene sulfides**
69 5,326,362 **Method of surface hardening orthopedic implant devices**
70 5,316,863 **Self-brazing aluminum laminated structure**
71 5,314,672 **Composition and method for sweetening hydrocarbons**
72 5,290,368 **Process for producing crack-free nitride-hardened surface on titanium by laser beams**
73 5,248,386 **Milling solution and method**
74 5,233,060 **Ethylene recovery in direct-oxidation ethylene oxide processes**
75 5,226,981 **Method of manufacturing corrosion resistant tubing from welded stock of titanium or titanium base alloy**
76 5,215,624 **Milling solution and method**
77 5,203,982 **Cation exchange membranes**
78 5,198,308 **Titanium porous surface bonded to a cobalt-based alloy substrate in an orthopaedic implant device**
79 5,152,795 **Surgical implants and method**
80 5,124,528 **Gas tungsten and plasma arc welding electrode having a carbide emitter end**
81 5,100,500 **Milling solution and method**
82 5,092,760 **Oxygen-fuel burner assembly and operation**
83 5,073,356 **Integrated processes for the production of carbon monoxide**
84 5,057,133 **Thermally efficient melting and fuel reforming for glass making**
85 5,006,141 **Thermally efficient melting for glass making**
86 5,004,420 **Dental bridge and manner for preparation**
87 4,999,051 **System and method for atomizing a titanium-based material**
88 4,987,033 **Impact resistant clad composite armor and method for forming such armor**
89 4,980,127 **Oxidation resistant titanium-base alloy**
90 4,954,708 **Low distortion focal plane platform**
91 4,899,670 **Means for providing oxygen enrichment for slurry and liquid fuel burners**
92 4,886,651 **Process for co-production of higher alcohols, methanol and ammonia**
93 4,876,065 **Corrosion-resistant Fe-Ni-Cr alloy**
94 4,869,741 **Ultra pure liquid oxygen cycle**
95 4,863,474 **Skeletal implants**
96 4,854,496 **Porous metal coated implant and method for producing same**
97 4,836,723 **Spindle assembly**
98 4,834,919 **Stereoselective reduction of the keto group at 7-position of a bile keto acid**
99 4,717,068 **Process for plating Al alloys containing Li, by hot co-rolling**
100 4,712,550 **Retinal tack**

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PAT. NO. Title

- 101 4,708,860 Split-flow regeneration in absorptive air separation
- 102 4,695,318 Method of making steel
- 103 4,687,512 Desulfurizing mixture for metal melts, process for making it, and process for desulfurizing liquid metal therewith
- 104 4,673,967 Surface mounted system for leaded semiconductor devices
- 105 4,648,901 Introducing one or more metals into a melt comprising aluminum
- 106 4,645,453 Bendable adapter for dental implant
- 107 4,636,219 Prosthesis device fabrication
- 108 4,608,227 Sintered titanium horseshoes
- 109 4,589,960 Activated metal anodes and a process for making them
- 110 4,581,064 Treatment of anode slimes in a top blown rotary converter
- 111 4,565,685 Air separation with temperature and pressure swing
- 112 4,564,393 Introducing one or more metals into a melt comprising aluminum
- 113 4,559,089 Method for making a light weight composite of pure aluminum, heat treatable aluminum, and stainless steel
- 114 4,544,116 Helicopter landing skid shoe pad
- 115 4,542,114 Process for the recovery and recycle of effluent gas from the regeneration of particulate matter with oxygen and carbon dioxide
- 116 4,526,775 Oxygen production by molten alkali metal salts using multiple absorption-desorption cycles
- 117 4,519,401 Pressure telemetry implant

118 4,464,442 **Brazing alloy**
119 4,408,982 **Process for firing a furnace**
120 4,313,759 **Wear resistant aluminium alloy**
121 4,216,089 **Waste water treatment**
122 4,201,902 **Electrode for air-carbon arc cutting and gouging**
123 4,180,547 **Process for recovery of chemicals from saline water**
124 4,179,287 **Method for adding manganese to a molten magnesium bath**
125 4,178,174 **Direct production of copper metal**
126 4,170,326 **Method and holding fixture for soldering lead frames to hybrid substrates**
127 4,165,979 **Flash smelting in confined space**
128 4,156,268 **Humidity sensing element and method of manufacture thereof**
129 4,148,630 **Direct production of copper metal**
130 4,137,032 **Corrosion-resistant spinneret**
131 4,129,438 **Method of adding trace elements to base metals**
132 4,125,924 **Method of producing composite metal pipe**
133 4,119,701 **NO_x reduction catalyst for internal combustion engine emission control**
134 4,118,291 **Method of electrowinning titanium**
135 4,116,801 **Apparatus for electrowinning multivalent metals**
136 4,110,180 **Process for electrolysis of bromide containing electrolytes**
137 4,085,997 **Anodize clamp**
138 4,072,511 **Method of producing silicon containing cast iron**
139 4,064,221 **Process for obtaining nitric acid of a concentration higher than the azeotropic concentration by means of the absorption of nitrogen oxides in water or diluted nitric acid**
140 4,054,468 **Process of making a laminated spinneret**
141 4,006,105 **NO_x reduction catalyst for internal combustion engine emission control**
142 3,984,231 **Process for the two-stage reduction of iron ore in a rotary kiln**
143 3,960,546 **Method for eliminating nose-skulls from steelmaking vessels**

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